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Matter All Around Us

Textbook pages 190-194 | Lesson

1

 Read 10.1 “Matter All Around Us” on pages 190-194 of the textbook.



Exploring the Lesson

A Write the letter of each vocabulary word beside its definition.

1. ____ to connect or stick tightly together
2. ____ the tiniest particle of matter
3. ____ a particle made of two or more atoms bonded together
4. ____ something that takes up space and has weight

a. atom
b. bond
c. matter
d. molecule

B Circle the letter of the answer to each question.

5. How do we know that air is matter?
 - a. Air can be either hot or cold.
 - b. Air takes up space and has weight.
 - c. The wind is moving air in the atmosphere.
6. How do we know that light is *not* matter?
 - a. A flashlight can shine into a box.
 - b. Light cannot be weighed.
 - c. The sun is a source of light.
7. Which set of things is made of matter?
 - a. electricity, heat, sound
 - b. light, love, laughter
 - c. metal, rocks, water



Lesson 1



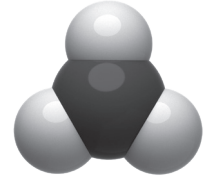
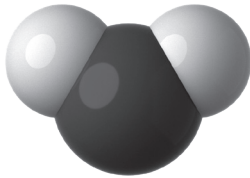
C Underline the bold word that completes each sentence.

8. All matter is made of **air, atoms**.
9. An atom **can, cannot** be cut or made smaller.
10. Air contains atoms of oxygen and **aluminum, nitrogen**.
11. A water molecule contains two **carbon, hydrogen** atoms and one oxygen atom.

D Circle *T* if the statement is *true* or *F* if it is *false*.

12. **T** **F** Atoms can be seen with binoculars.
13. **T** **F** A drop of water contains billions of molecules.
14. **T** **F** Most matter is made of atoms that have formed molecules.
15. **T** **F** Atoms can join together in only one way.

E Write *atom* or *molecule* to label each drawing.



16. _____ 17. _____ 18. _____

F Write the type of matter that would be used to make each set of things.

concrete cotton metal wood

19. _____ socks, blankets, pillows
20. _____ furniture, cabinets, benches
21. _____ vehicles, tools, wire
22. _____ foundations, driveways, sidewalks

The Three Forms of Matter

Textbook pages 195-198 | Lesson **2**

 Read 10.2 “The Three Forms of Matter” on pages 195-198 of the textbook.



Exploring the Lesson

A Complete the definition of each vocabulary word.

1. **gas:** matter with _____ definite _____ or definite _____
2. **liquid:** matter with a definite _____ but _____ definite _____
3. **solid:** matter with _____ a definite volume and a definite shape
4. **volume:** the amount of _____ matter takes up

B Write *gas*, *liquid*, or *solid* to label the three forms of water.



5. _____

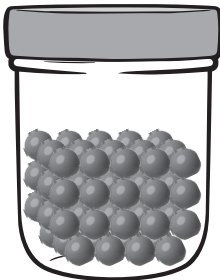


6. _____

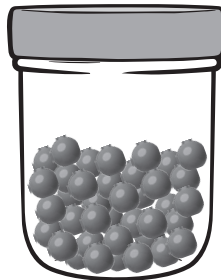


7. _____

C Write *gas*, *liquid*, or *solid* to label the atoms in the three forms of matter.



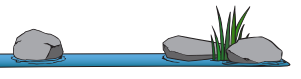
8. _____



9. _____



10. _____



D Place a check mark beside the answers to each question.

11. What two things happen to wooden blocks if you dump them out of a box?
- a. ____ The blocks melt and spread into a puddle on the floor.
 - b. ____ The blocks still have the same volume as they did in the box.
 - c. ____ The blocks keep their shape but are scattered.
12. What two things happen to liquid tea when it is poured into cups?
- a. ____ The tea still has the same volume.
 - b. ____ The tea takes the shape of the cups.
 - c. ____ The tea goes into the air, and you cannot see it.
13. What two things happen when gas from a helium balloon is released into a small room?
- a. ____ The helium forms drops of liquid in the air.
 - b. ____ The helium molecules spread throughout the room.
 - c. ____ The helium takes the shape of the room instead of the shape of the balloon.
14. Which three items describe a solid that has been changed but is still a solid?
- a. ____ a tree chopped into firewood
 - b. ____ a block of ice that has melted into water
 - c. ____ a cake cut and shaped to look like a train
 - d. ____ kernels of wheat that have been ground into flour

E Write the letter of each form of matter beside its examples. One answer will be used twice.

15. ____ apple, car, rock
16. ____ blood, juice, milk
17. ____ carbon dioxide, helium, hydrogen
18. ____ ice, rubber tire, blanket

- a. gas
b. liquid
c. solid



We Remember

F Complete each sentence.

19. All matter takes up _____ and has _____ .
20. All matter is made of _____ .
21. An atom is the _____ particle of matter.
22. Atoms form _____ when they bond together.
23. An _____ cannot be cut or made _____ .

G Place a check mark beside the answers to each question.

24. Which five things are made of matter?

- | | | |
|---------------------|----------------|---------------|
| a. ____ air | d. ____ heat | g. ____ rock |
| b. ____ balloon | e. ____ light | h. ____ sound |
| c. ____ electricity | f. ____ pencil | i. ____ water |

25. Which two types of atoms form water molecules?

- | | | |
|----------------|------------------|----------------|
| a. ____ carbon | b. ____ hydrogen | c. ____ oxygen |
|----------------|------------------|----------------|

H Draw diagrams to illustrate an atom and a molecule.

26. Atom

27. Molecule

Matter Changes

Textbook pages 199-203

Lesson

3



Read 10.3 “Matter Changes” on pages 199-203 of the textbook.



Exploring the Lesson

A Write the word that completes each vocabulary definition.

- _____ **change:** a change to the shape or form of matter
- _____ **change:** a change that produces a different substance

B Place a check mark beside the answers to the question.

3. Which six actions show a physical change?

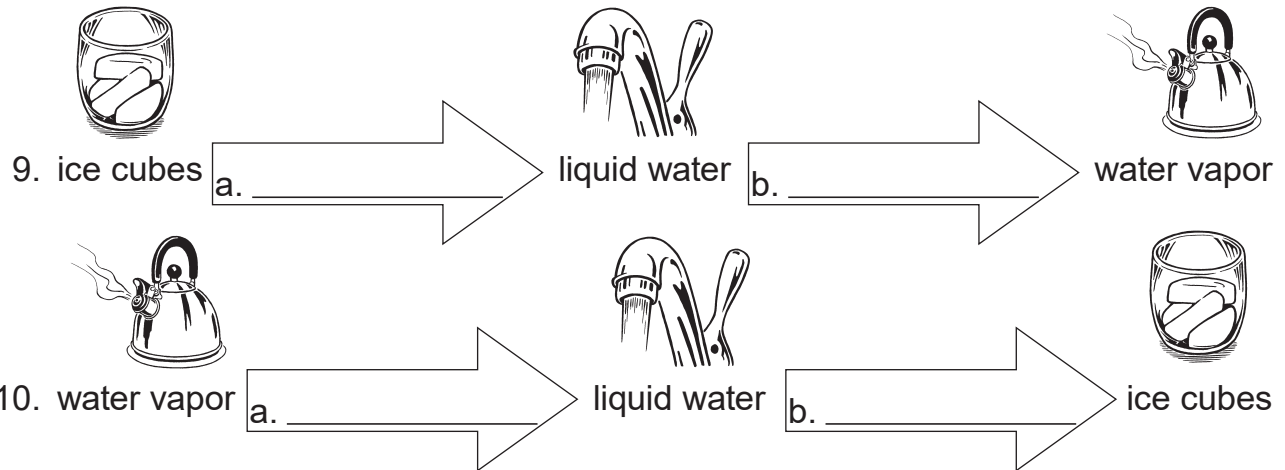
- | | |
|---|--|
| a. _____ bending a wire | e. _____ baking bread |
| b. _____ burning a piece of paper | f. _____ shaping clay into a bowl |
| c. _____ mixing soil with sand | g. _____ cutting fabric |
| d. _____ splitting a log into
firewood | h. _____ melting an ice cube into
water |

C Write the letter of what results from each chemical change.

- _____ burning firewood
- _____ frying an egg
- _____ decaying leaves
- _____ burning gasoline
- _____ combining iron and oxygen

- compost
- gases and water vapor
- hardened proteins and fats
- rust
- gases and a pile of ash

D Write *cooling* or *heating* to show how matter changes form.



E Underline the bold word that completes each sentence.

- A **chemical, physical** change produces a different substance.
- Matter can change form, but its **atoms, shapes** are not destroyed.
- When a log burns, the atoms in the wood **rearrange, disappear** to form other substances.
- Although changes occur, the **amount, shape** of matter is always the same.



We Remember

F Write the definition of the vocabulary word.

15. **volume:** _____

G Write *yes* or *no* to identify each characteristic of the forms of matter.

	Solid	Liquid	Gas
Definite volume	16. _____	18. _____	20. _____
Definite shape	17. _____	19. _____	21. _____



Lesson 3



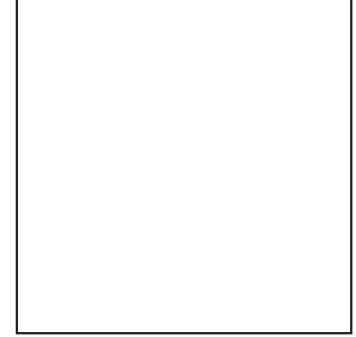
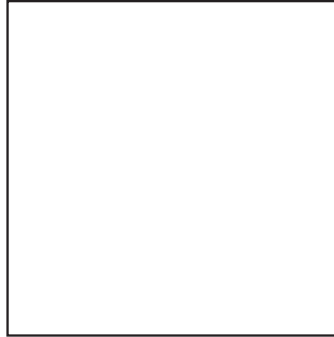
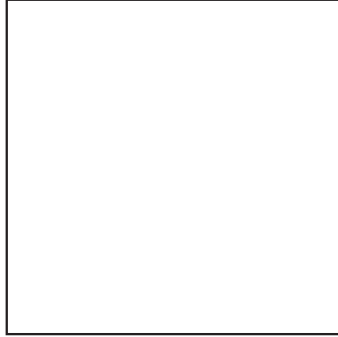
- **H** Draw diagrams to illustrate the atoms in a solid, a liquid, and a gas. Use the diagrams in Lesson 2 for a model if needed.

22.

a. Solid

b. Liquid

c. Gas



- **I** Cross out the substance that does not belong in each group.

23. **solid**: ice, juice, pencil, doorknob

24. **liquid**: clay, rain, diesel fuel, hot tea

25. **gas**: helium, nitrogen, oxygen, plastic

- **J** Underline the bold word that completes each sentence.

26. An atom is the **tiniest, largest** particle of matter.

27. When atoms bond, they form **molecules, liquids**.

28. All matter takes up space and has **shape, weight**.

29. All matter is made of **atoms, air**.

30. Air, metal, water, and rocks are all made of **solids, matter**.



Investigation: Make Ice Cream

Lesson **4**

How Does Matter Change Form?

Matter changes are all around us. In this investigation, you will observe matter changing from a liquid mixture into a solid as you make ice cream.

Materials

You will need the following items:

- $\frac{1}{2}$ cup half and half, heavy whipping cream, or milk
- 1 tbsp. sugar
- $\frac{1}{2}$ tsp. vanilla extract
- toppings of your choice
- 3 cups of ice cubes
- $\frac{1}{3}$ cup salt (preferably rock salt or kosher salt)
- at least one sealable plastic sandwich bag
- one gallon-size Ziploc bag
- oven mitts or a small towel